

Free radical scavenging activity of methanolic and aqueous extract of *Albizia lebbeck* leaves

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ABSTRACT

Antioxidants are emerging as prophylactic and therapeutic agents for various diseases. The present study deals with an antioxidant potential of methanolic and aqueous extract of *Albizia lebbeck* leaves. The antioxidant potential of methanolic and aqueous extract of *A. lebbeck* leaves was investigated by employing two *in vitro* methods, such as Nitric oxide radical scavenging activity and reducing power assay. Results obtained in the present study reveal methanolic extract of *A. lebbeck* possesses significant antioxidant activity as compared to aqueous extract.

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Keywords: *Albizia lebbeck* leaves, antioxidant, nitric oxide radical scavenging, reducing power assay etc.

Introduction

Ever since the birth of humanity, there has been a relationship between life, disease, and plants. Primitive men started studying diseases and treatments. There is no record that people in prehistoric times used synthetic medicines for their ailments but they tried to make use of the things they could easily procure. The most common thing they could find was there in environment, i.e., the plants and animals. They started using plants and found that majority of plants were suitable as food, whereas other were either poisonous or medicinally useful. By their experience, this knowledge of herbal remedies was transferred to generation as folk medicine. Hence, the history of herbal medicine is as old as human history. Herbal medicine is still the mainstay of about 75-80% of the world's population, mainly in the developing countries, for primary health care because of better cultural acceptability, better compatibility with the human body, and lesser side effects. It is estimated that approximately one-quarter of prescribed drugs contain plant extracts or active ingredients obtained from or modeled on plant substances.^[1]

Higher plants have been used as a source of drugs by humanity for several 1000 years. In fact, ancient man was totally dependent on green plants for his day-to-day needs of medicaments. With the development of modern medicine, synthetic drugs and antibiotics, the importance of plants as raw material for drugs decreased considerably. However, plants were used as a basis of some of the most important drugs, even in the modern system of medicine. With the advancement of synthetic organic chemistry, most of the active constituents of plants used in medicine were synthesized. At one time, it was thought that ultimately all the plant drugs would be obtained from synthetic sources. However, in spite of phenomenal progress in the development of new drugs from synthetic sources and the appearance of antibiotics as major therapeutic agents, plants continue to provide basic raw materials for some of the most important drugs.^[2]

Today in many countries, modern medicine has displaced plants with many synthetic products but almost 30% of pharmaceutical preparations are still obtained directly or indirectly from plants. The modern era has seen some decline in the use of medicinal plants and their extracts as therapeutic agent, particularly in developed countries, many of which either been discarded by the medical profession or now given in the form of isolated compound.^[3]

Antioxidant means "against oxidation" and the work to protect lipid from peroxidation by radicals. The human body as an elaborate antioxidant defense system. The main characteristics of an antioxidant are its ability to trap free radicals. Highly reactive free radicals and oxygen species are present in biological system from a wide variety of sources. These free radicals may oxidize nucleic acid, proteins, lipids, or DNA and can initiate degenerative disease. An antioxidant compound like phenolic acids, polyphenols, and flavonoid scavenge

Access this article online

Website: <http://www.jpbs-online.com>

E-ISSN: 2321-0125

DOI: 10.31555/jpbs/2018/6/1/9-12

How to cite this article: Umesh PJ, Swapnesh AJ, Kailaspati PC, Minal SP, Rajendra DW. Free radical scavenging activity of methanolic and aqueous extract of *Albizia lebbeck* leaves. J Pharm BioSci 2018;6(1):9-12.

Source of Support: Nil, **Conflict of Interest:** None declared.

free radicals such as peroxide, hydroperoxide, or lipid peroxyl and thus inhibit the oxidative mechanism that leads to degenerative diseases. There are a number of clinical studies suggesting that the antioxidant in fruits, vegetables, tea, and red wine are the main factors for the observed efficacy of these foods in reducing the incidence of chronic diseases including heart disease and some cancers.^[4]

Free radicals are defined as any atom or molecule having unpaired electrons. They are involved in various physiological processes (inhibition of peroxidation of the membrane lipids, inhibition of mitochondrial respiratory chain enzymes, etc.) and play a major role in the inception of many diseases and ultimately lead to cell death. They also affect the food's sensory quality-color, taste, and texture, which also shorten the shelf life and can result in rejection on the part of consumers. Free radicals are implicated for many diseases including diabetes mellitus, arthritis, cancer, and aging, in the treatment of these diseases antioxidant therapy has gained utmost importance.^[5]

Many herbal remedies individually or in combination have been recommended in various medical expositions for the cure of different diseases. Plant product serves the best source for controlling these activities by its own metabolic pathway. Currently, there has been an increased interest globally to identify an antioxidant compound that is pharmacologically potent and has low or no side effects. As plants are a source of natural antioxidants, much attention has been given to plants. A variety of free radicals scavenging antioxidants exists within the body in which many of them are derived from dietary sources such as fruits, vegetables, and teas. Dietary antioxidants are defined as a substance in foods that significantly decreases the adverse effect of reactive species, such as reactive oxygen and nitrogen species, on normal physiological function in humans. Natural antioxidants have gained increasing interest among consumers and the scientific community because they are economic, does not reduce food quality and do not exert side-effects like synthetic antioxidants. The antioxidant property may be related to the phenols and flavonoids present in the extracts.^[6]

Material and Methods

Collection and identification of plant material

Albizia lebbek leaves were collected from local area of District-Dhule, Maharashtra, India. The identity of the plant material was verified by Dr. D. A. Dhale, Head, SSVPS's Science College, Dhule, Maharashtra, India. A voucher specimen (No-1) is deposited in the herbarium of SSVPS's Science College, Dhule, Maharashtra, India. The plant material was dried in a hot air oven (<50°C), stored in airtight glass bottles and powdered to 40 mesh.

Extraction

The dried powdered plant material was submerged in sufficient volume of methanol and aqueous in an air-tight flat bottomed container for 7 days, with occasional shaking and stirring. The extracts were then filtered evaporated on electrical water bath and dried in vacuum oven.

Chemicals

Sodium nitroprusside solution, sulphanilamide, O-phosphoric acid, naphthyl ethylenediamine dihydrochloride, standard phosphate buffer, ascorbic acid, potassium ferricyanide, trichloroacetic acid, ferric chloride solution (0.1% w/v).

In vitro antioxidant evaluation

- Nitric oxide (NO) radical scavenging activity^[7,8]
Prepare different concentrations of extract in standard phosphate buffer in concentration range 25, 50, 75, 100, 125, 150, 175, and 200 µg/ml. Mix 1ml sodium nitroprusside (10 mM) in phosphate buffered saline with 1 ml of each extract and standard ascorbic acid separately and incubate at room temperature for 150 min. The same reaction mixture without the extract sample but with an equivalent amount of standard phosphate buffer should be serving as control. After the incubation period, to 1.5 ml of above reaction mixture; add 0.5 ml of Griess reagent. Immediately measure the absorbance of the chromophore formed at 546 nm. Determine the percentage scavenging activity at different concentrations. Carry out the same assay in triplicate.
- Reducing power assay.^[9,10]
The reducing power of methanolic and aqueous extract was determined. Different concentration of methanolic and aqueous extract (25, 50, 75, ... 200 µl/ml) was prepared and mix 1.0 ml of each sample with 2.5 ml of phosphate buffer solution (50 mM, pH 7.0) and 2.5 ml of 1% potassium ferric cyanide separately, incubate at 50°C for 20 min. Then add 2.5 ml of trichloroacetic acid (10%) to the mixture, centrifuge at 3000 rpm for 10 min. Finally, mix 1.25 ml from supernatant with 1.25 ml of distilled water and 0.25 ml of FeCl₃ solution (0.1% w/v). Immediately measure the absorbance at 700 nm. Carry out the same assay in triplicate. Prepare the different concentration of sample results of this method is based on the fact of increased absorbance value of different concentrations indicates higher reducing power.

Statistical analysis

The data presented as Mean ± SEM. The activities of all extracts were compared with the control. All the extracts showed significant activity with a higher duration of paralysis and death. Values of $P < 0.001$ were considered statistically significant.

Result and Discussion

In Nitric oxide radical scavenging activity model, it is observed that methanolic and aqueous extract of leaves of *A. lebbek* have demonstrated dose dependent increase in the nitric oxide anion scavenging property. The methanolic extract shows potential antioxidant activity as compared to aqueous extract shows 88.17 ± 0.62 and 79.19 ± 0.62 percentage inhibition at 200 µg/ml respectively and ascorbic acid has shown 95.26 ± 0.24 percentage inhibition at 200 µg/ml as shown in Table 1. The IC₅₀ value for methanolic and aqueous extract and standard ascorbic acid as shown in Table 2.

Table 1: Nitric oxide radical scavenging activity of methanolic and aqueous extract of plant *Albizia lebbek* with standard ascorbic acid (percentage inhibition)

Concentration (µg/ml)	Methanolic extract (% inhibition)	Aqueous extract (% inhibition)	Ascorbic acid (% inhibition)
25	29.54±0.62*	23.16±0.62*	38.76±0.85*
50	39.94±0.47*	30.49±1.08*	52.24±2.25*
75	48.69±0.62*	40.18±0.23*	59.09±0.62*
100	58.38±0.23*	49.16±0.85*	64.76±0.62*
125	65.71±0.62*	57.67±0.85*	71.17±0.63*
150	73.27±0.23*	65.47±0.85*	81.79±0.62*
175	79.90±0.62*	72.57±1.24*	89.59±0.63*
200	88.17±0.62*	79.19±0.62*	95.26±0.24*

*Data and results are expressed as mean±SEM and mean is representation of three experiments, *Albizia lebbek*: *A. lebbek*

Table 2: IC₅₀ result nitric oxide radical scavenging activity of various extracts of leaves of *Albizia lebbek* with standard ascorbic acid

Extract	IC ₅₀
Methanolic	101.21 µg/ml
Aqueous	114.68 µg/ml
Ascorbic acid	90.58 µg/ml

Calculated by regression equation, *Albizia lebbek*: *A. lebbek*

Table 3: Observation of reducing power determination of ascorbic acid, methanolic, and aqueous extract of leaves of *Albizia lebbek*

Concentration (µg/ml)	Absorbance (nm)		
	Ascorbic acid	Methanolic extract	Aqueous extract
25	0.035±0.0008*	0.029±0.0847*	0.028±0.0010*
50	0.051±0.0005*	0.047±0.0008*	0.037±0.0011*
75	0.069±0.0005*	0.058±0.0006*	0.043±0.0003*
100	0.077±0.0008*	0.071±0.0003*	0.052±0.0003*
125	0.090±0.0006*	0.075±0.0003*	0.064±0.0008*
150	0.099±0.0008*	0.092±0.0006*	0.078±0.0003*
175	0.136±0.0008*	0.096±0.0003*	0.082±0.0008*
200	0.171±0.0011*	0.108±0.0008*	0.091±0.0011*

*Data and results are expressed as mean±SEM and mean is representation of three experiments, *Albizia lebbek*: *A. lebbek*

It is observed that the both methanolic and aqueous extract of plant *A. lebbek* demonstrated dose dependant increase in the reducing property. To find the active species which is capable of donating hydrogen and subsequently its leads to the reducing power activity was determine. The high reducing power is indicative of the hydrogen donating ability of the active species present in extract. The reducing power assay of various extract of plant *A. lebbek* was estimated by using potassium ferricyanide reduction method. In the present study the reducing power of the methanolic extract of plant *A. lebbek* was found to be excellent and steadily increase in direct proportional to the increasing concentration extract as compare to other extract in comparison with standard ascorbic acid. The reducing power of methanolic, aqueous and standard ascorbic acid at concentration 200 µg/ml was found to be 0.108, 0.091 and 0.171 respectively as shown in Table 3.

Nitric oxide is an important chemical mediator generated by endothelial cells macrophages, neurons and is involved in regulation of various physiological processes. Excess concentration of nitric oxide is associated with several diseases. Nitric oxide is generated in biological tissues by specific nitric oxide synthesis (NOS) which metabolized arginine to adrenaline with formation of nitric oxide via a five electron oxidative reaction. These compounds are responsible for altering the structural and functional behavior of many ocular components.

The percentage scavenging activity increased with increasing concentration of the extract. Lower the IC₅₀ value indicates better is the scavenging ability of the sample. The nitric oxide scavenging activity of methanol extract showed better activity than other extract, however standard ascorbic acid activity was significantly higher than that of all extract. The reducing power of extract of *A. lebbek* was found remarkable and the reducing power of the extract was observed to rise as the concentration of the extract gradually increased.

The reducing power is associated with antioxidant activity and many serve as a significant reflection of the antioxidant activity. Compound with reducing power indicate that they are electron donors and can reduce the oxidized intermediates of lipid preoxidation processes, so that they can act as primary and secondary antioxidants. In this assay the yellow color of test solution changes to various shades of green and blue depending on the reducing power of each compound, presence of reducers causes the conversion of Fe³⁺/ferricyanide complex used in this method to ferrous form. The high reducing power is indicative of the hydrogen donating ability of the active species present in the extract.

Reducing power of the fractions was assessed using ferric to ferrous reducing activity as determined spectrophotometrically from formation of blue color complex. Reducing power of different extract of plant *A. lebbek* was compared with standard ascorbic acid. Methanolic extract exhibited most reducing power. This extract indicates that extract may consist of polyphenolic compounds that usually shows great reducing power.

Conclusion

We conclude from the above discussion that those methanolic extracts have antioxidant activity by scavenging the nitric oxide free radical. It is very much helpful for investigation of new drugs for various free radical generation diseases by identifying the compound isolation process.

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